## Versatile starting materials for novel 1, $\omega$ -bis(pyridin-4-ylphenoxy)alkanes, and their corresponding bis(thieno[2,3-b]pyridin-4-ylphenoxy) derivatives Ashraf A. Abbas,\*Mohamed A.A. Elneairy and Yehia N. Mabkhot

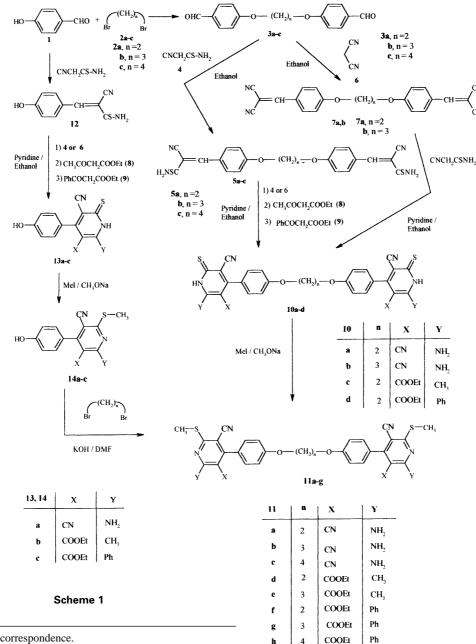
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A synthesis is described, starting from *p*-hydroxybenzaldehyde, of some new bis(activated styrene) derivatives, and their conversion into novel bis(pyridin-4-yl) ethers and bis(thieno[2,3-*b*]pyridine) derivatives.

Keywords: 1,@-bis(pyridin-4-ylphenoxy)alkanes, bis(thieno[2,3-b] pyridin-4-ylphenoxy) derivatives

In recent decades considerable attention has been devoted to the construction of new derivatives of pyridine-2-thiones and thieno[2,3-*b*]pyridine on account of their reported biological activities. Various series of substituted pyridine-2-thiones have been found to be useful as antibiotic<sup>1-2</sup>, antiarteriosclerotic<sup>3</sup>, antibacterial<sup>4</sup>, antihyperglycemic<sup>5</sup> and antifungal<sup>6</sup> agents and as inhibitors of the blood coagulation factor.<sup>7</sup> Also, thieno[2,3-*b*]pyridine derivatives have been reported to be useful as antibiotics<sup>8-11</sup>, drug intermediates<sup>12</sup>, against endoparasiticides<sup>13</sup> and as antianaphylactic compounds.<sup>14-15</sup> Recently bis(compounds) have received great attention, not only for being model compounds for main chain poly-



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mers,<sup>20–25</sup> but also because many biologically active natural and synthetic products have molecular symmetry.<sup>26</sup> We have recently described the synthesis of some new bis( $\beta$ -difunctional) building units and studied their use as intermediates in the synthesis of novel bis(5-membered heterocycles).<sup>27</sup> In continuation of our interest in this field, this paper describes the synthesis of some new bis(activated styrene) derivatives to explore their synthetic potential as starting materials for novel bis(pyridin-4-yl) ethers and bis(thieno[2,3-*b*]pyridine) derivatives and their utility as pharmacological agents.

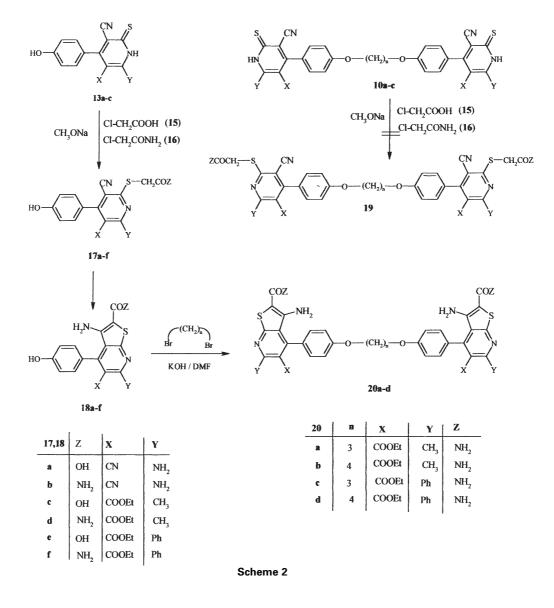
## **Results and discussion**

Two synthetic routes are described for the synthesis of the bis(2-methylthiopyridin-4-yl) ethers **11a-h** as outlined in Scheme 1. The starting compounds **5a-c** and **7a**, **b** needed in the present investigation were prepared by condensing the appropriate bis(aldehydes) **3a-c**<sup>28</sup> with 2-cyanothioacetamide (**4**) and malononitrile (**6**) in absolute ethanol containing triethylamine (TEA) as catalyst. Compounds **5a-c** have been reacted with a series of active methylene compounds, namely 2-cyanothioacetamide (**4**), malononitrile (**6**), ethyl acetoacetate (**8**) and ethyl benzoylacetate (**9**) in refluxing ethanolpyridine mixture to give 35–40% of the corresponding bis(pyridin-4-yl) ethers **10a-d**. The latter were methylated to yield 50–60% of the corresponding bis(2-methylthio) deriva-

tives **11a**, **b**, **d**, **f**. Compounds **10a**, **b** were alternatively obtained in 45% yield by reacting **7a**, **b** with **2** in refluxing ethanol-pyridine mixture.

In view of the low yields of the bis(pyridin-4-yl) ethers 10a-d and their corresponding bis(methylthio) derivatives 11a, b, d, f, we developed a second route (Scheme 1) for the preparation of 11a-h. The pyridines 14a-c were the key intermediates; their potassium salts (obtained from 14a-c with KOH/MeOH) reacted with the appropriate dibromoalkanes 2b, c in boiling DMF to give 60–68% yield of 11b, c, e, g, h. Compounds 14a-c were obtained in 70-75% yield by methylation of 13a-c with methyl iodide in refluxing methanol containing sodium methoxide. Compounds 14a-c were confirmed as the S-methyl derivatives by the IR and <sup>1</sup>H NMR spectra. Thus, the IR spectra of these compounds showed the absorption bands of phenolic OH at ca. 3330 cm<sup>-1</sup>. Also their <sup>1</sup>H NMR spectra revealed singlet signals at  $\delta \cong 2.60$  and  $\delta \cong$ 10.2 ppm, characteristic for SCH<sub>3</sub> and phenolic OH protons respectively.

Compounds **13a–c** were obtained by first condensation of 4-hydroxybenzaldehyde (**1**) with 2-cyanothioacetamide (**4**) in absolute ethanol containing TEA to give 2-cyano-3-(4-hydroxyphenyl)thioacrylamide (**12**).<sup>29</sup> Subsequent reaction of **12** with each of **4**, **6**, **8**, **9** in refluxing ethanol containing TEA afforded the corresponding pyridine-2-thiones **13a–c**. The lat-



ter were shown to exist mainly in the thione form by the strong absorption bands at  $v \approx 3300$  and 1540 cm<sup>-1</sup> characteristic for NH and C=S groups respectively.

With the pyridine-2-thiones 13a-c and the bis(pyridin-4-yl) ethers 10a-c now available, we attempted two routes to the synthesis of the bis(thieno[2,3-*b*]pyridin-4-yl) ethers 20a-d, as outlined in Scheme 2. In the first route we tried alkylation of 10a-c with chloracetic acid (15) or chloroacetamide (16) in methanolic sodium methoxide to give the corresponding bis-S-alkyl derivatives 19. The latter would be expected to undergo cyclization in ethanolic potassium hydroxide to afford the target molecules 20. Unfortunately we could not isolate pure samples of the alkyl products 19 under the conditions described.

In the second route the pyridine-2-thiones **13a–c** were easily converted into the corresponding *S*-(2-pyridyl)thioglycolic acids **17a, c, e** and 2-pyridylthioacetamides **17b, d, f** upon treatment with chloracetic acid (**15**) and chloroacetamide (**16**), respectively, in refluxing methanol containing sodium methoxide. Compounds **17a–f** could be cyclized to give the thieno[2,3-*b*]pyridine derivatives **18a–f** in 60-70% yields upon heating in ethanol with potassium hydroxide. The potassium salts of **18c-f** reacted with 1,3-dibromopropane (**2b**) and 1,4-dibromobutane (**2c**), respectively, in boiling DMF to furnish the corresponding bis(thieno[2,3-*b*]pyridine) derivatives **20a–d** in 60-65 % yield.

The substituted thieno[2,3-b]pyridine derivatives **18a**, **b** could not be converted into the corresponding bis(thieno[2,3-b]pyridin-4-yl) ethers under similar conditions. This may be attributed to the insolubility of these derivatives under the reaction conditions.

Techniques used: Elemental analysis, IR, <sup>1</sup>H NMR and MS.

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